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Growth Conditions of Sequentially Electrodeposited Buffer Layers for YBCO Superconductor

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Abstract— We report recent technical advances in the fabrication by sequential electrodeposition (ED) of buffer architectures with two stacking layers. The proposed approach is to fabricate a structure containing a pyrochlore oxide and Gd₂O₃, which takes advantage of the low oxygen diffusivity in pyrochlores and the excellent crystal lattice match of Gd₂O₃ and YBa₂Cu₃O₇₋₈ (YBCO). A systematic study is performed for a fundamental understanding of the buffer structure grown by ED. composition, electrolyte solution composition/concentration of chemicals, conditions are found to affect considerably the deposition of the layer, formation of the pyrochlore oxide, and film texture.

Index Terms— Buffer layers, Electrodeposition, Pyrochlore compound, RE_2O_3 compound, Texture.

I. INTRODUCTION

FOR several decades, intensive research and development efforts have been carried out to integrate superconductor films into various devices such as cables, transformers, current limiters, magnets, and motors. YBCO films prepared on rolling-assisted biaxially textured substrates (RABiTS) [1] and also YBCO films prepared by ion-beam assisted deposition (IBAD) [2] on non-textured substrates have shown promising performance with MA/cm² current density at 77 K [3]. The RABiTS process is based on the deposition of an epitaxial oxide buffer layer on a textured metal substrate. The buffer layer structure, a stack of at least three oxide layers, functions as a template for the growth of the YBCO superconducting layer and prevents metal and oxygen diffusions between the substrate and the superconductor.

To produce coated conductors in lengths of 100 meters or more, some important issues (such as slow production rate, high cost, and inhomogeneity in long tapes) need to be resolved in these thin-film technologies [4]. Our approach to the fabrication of coated conductors is based on non-vacuum techniques; in particular, electrodeposition has potential [5]

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for fabricating long-length, high-rate, low-cost tapes.

Among the materials compatible with YBCO, oxides with structures of perovskite (LaMnO₃), fluorite (CeO₂), pyrochlore (Gd₂Zr₂O₇), and REO (Gd₂O₃) have shown promising properties as oxygen barriers. We have focused our efforts on two simplified buffer architectures: Gd₂O₃/Gd₂Zr₂O₇ and CeO₂/Gd₂Zr₂O₇ on a RABiTS Ni-W substrate, which takes advantage of the low oxygen diffusivity in pyrochlores and the excellent crystal lattice match of CeO₂ and Gd₂O₃ with YBCO. The processing steps are as follows:

- 1. Electrodeposition of Gd₂Zr₂O₇ layer on Ni-W substrate
- 2. Electrodeposition of Gd₂O₃ or CeO₂ on Gd₂Zr₂O₇ layer
- 3. Single ex-situ thermal treatment.

Buffer layers with optimal physical properties and morphology are indispensable for the growth of highly superconducting thin film. Texture and surface morphology in particular are crucial for the application of the films in planar multilayer devices such as coated conductors. We obtained very high-quality electrodeposited buffer layers by optimizing the electrolyte composition at a fixed deposition current density and also optimizing the subsequent thermal treatments. In this paper, we report on the effect of the electrolyte mixture and the thermal treatments on the formation of electrodeposited $Gd_2Zr_2O_7$ and $Gd_2O_3/\ Gd_2Zr_2O_7$ layers.

II. EXPERIMENTAL DETAILS

 $Gd_2Zr_2O_7$ films were electrodeposited from a bath containing 5 mM of gadolinium bromide and 2 mM of zirconium bromide dissolved in dimethyl sulfoxide (DMSO) and ethanol solution mixture. Electrodeposition was performed at an optimal current density of 1 mA/cm² and under constant stirring in a vertical two-electrode cell configuration. The average rate of deposition was about 30 nm/min. Gd_2O_3 films were electrodeposited using an aqueous bath under constant stirring and containing 6 mM of gadolinium bromide. The average rate of deposition was about 5 nm/min.

Composition and thickness of electrodeposited films were obtained from inductively coupled plasma (ICP) analysis. The annealed films were characterized by X-ray diffraction (XRD) using copper K α radiation ($\lambda=1.56054$ Å). Phase identification was performed in Bragg Bretano geometry ($\theta/2\theta$ geometry). The mosaic spread of $Gd_2Zr_2O_7$ grains along the c-

axis and in-plane texture were determined by rocking curve scans on the (004) reflection [2θ =33.90 (\pm 0.15)°] and pole figure scans on the (222) reflection [2θ =29.30 (\pm 0.03)°], respectively. The mosaic spread of Gd_2O_3 grains along the c-axis and in-plane texture were determined by rocking curve scans on the (002) reflection [2θ = 32.93 (\pm 0.03)°] and pole figure scans on the (111) reflection [2θ = 28.45 (\pm 0.03)°], respectively. The angular dispersion of $Gd_2Zr_2O_7$ and Gd_2O_3 grains in (a,b) plane was calculated from the peak broadening of ϕ -scans on the (222) reflection [$Gd_2Zr_2O_7$: 2θ =29.30 (\pm 0.03)° and χ =54.7°] and (111) reflections [Gd_2O_3 : 2θ = 28.45 (\pm 0.03)°], respectively. Morphology and roughness of films were revealed from scanning electron microscopy (SEM) scans and optical profiler scans.

III. RESULTS AND DISCUSSION

A. Effect of the electrolyte mixture on the electrodeposited $Gd_2Zr_2O_7$ layers

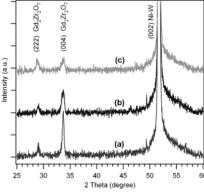


Fig. 1. X-ray diffraction after annealing at 950 °C of $Gd_2Zr_2O_7$ layers electrodeposited from electrolytes containing: (a) 6% ethanol, (b) 10% ethanol, and (c) 13% ethanol.

In this set of experiments, concentration of Gd³⁺ and Zr³⁺ ions were set to 5 and 2 mM, respectively. The films were deposited at a rate of 700 rpm with constant stirring by a magnetic stirring bar. We maintained a constant stirring speed to prevent inhomogeneities in the deposited films. Various electrolytes were selected for the deposition of Gd₂Zr₂O₇ layer on Ni-W substrate. The challenge in the electrodeposition of Gd₂Zr₂O₇ is to obtain crack-free and smooth surface morphology. A smooth and crack-free gadolinium oxide layer was easily electrodeposited from an aqueous media. On the other hand, the surface morphology of the zirconia layer electrodeposited from ZrOCl₂.xH₂O in aqueous media was poor. The layer deposited from a stagnant aqueous solution was reported to yield a rough surface with cracks [6,7]. Zirconia layers deposited from an organic solvent, DMSO, and alcohol were crack-free [8,9]. Thus a mixture of DMSO and ethanol was selected to electrodeposit the Gd₂Zr₂O₇ layer. The depositions were performed from various electrolytes containing 6%, 10%, and 13% ethanol in DMSO. Compositions of the electrodeposited layers Ga:Zr=1.27:0.73, Ga:Zr=1.28:0.72, approximately Ga:Zr=1.32:0.68, respectively. The thickness of the deposited layers was 75, 70, and 63 nm, respectively, when deposited using a 3-min deposition time for all films. ICP analysis revealed an increase of gadolinium in the layer and decreasing layer thickness when ethanol concentration was increased in the electrolyte. We also note that for electrolytes with high ethanol concentrations (>13%), a faster flow rate was necessary to prevent the formation of rough surface; this is probably due to the deposition of a gel-like layer containing saturated hydrocarbon, as has been observed in previous work [9]. As-deposited layers were amorphous, smooth, and crack free, as observed by an optical profiler. The layers were subsequently annealed at 950 °C for 1 to 5 h in an Ar:H2 gas mixture. The optimization of the gas mixture has been reported previously for the fabrication of textured ceria layers electrodeposited on Ni-W substrate [10]. A mixture of Ar:0.5%H₂ was selected for all processing temperatures, based on previous work.

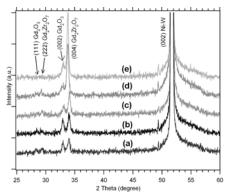


Fig. 2. X-ray diffraction of electrodeposited $Gd_2O_3/Gd_2Zr_2O_7$ on Ni-W substrate after annealing in Ar: H_2 gas flow at 1100 °C for (a) 1, (b) 2, (c) 3, (d) 4, and (e) 5 h.

XRD patterns of the three electrodeposited layers are shown in Figure 1. Patterns (a), (b), and (c) correspond to the layers electrodeposited in 6%, 10%, and 13% ethanol, respectively. Figure 1 shows three diffraction peaks at $2\theta = 28.83 \ (\pm 0.16)^{\circ}$, 33.66 (± 0.05)°, and 51.6 (± 0.03)°. The two peaks at 28.83° and 33.66° were identified as Gd₂Zr₂O₇ (222) and (004) reflections, respectively. The peak at $51.6 (\pm 0.03)^{\circ}$ is a Ni-W (002) reflection. In addition, Figure 1 shows a decreasing relative intensity of the Gd₂Zr₂O₇ (004) reflection in (b) and (c). As diffraction intensity is related to the crystallinity of the layer, the results suggest a decrease in crystallinity and grain growth for the layers electrodeposited in 10% and 13% ethanol. The increase of gadolinium content in the layer probably induces an amorphization of the layer. The best crystalline layer was obtained using 6% ethanol. However, misorientation along <111> axis was detected. The presence of (222) reflection is probably related to the non-stoichiometry of the layer as well as to the non-optimized annealing conditions. We optimized the annealing condition of the films obtained from 6% ethanol solution to eliminate the misorientation.

B. Ex-situ growth conditions for textured buffer stacking

Complete electrodeposited buffer architecture of the type $Gd_2Zr_2O_7/Gd_2O_3$ was deposited. About 60–70-nm-thick $Gd_2Zr_2O_7$ layers were deposited from a bath containing 6%

ethanol. A Gd₂O₃ cap layer about 30 nm thick was electrodeposited from an aqueous bath. The buffer layers were annealed at 1100 °C in H₂:Ar gas for 1 to 5 h. XRD patterns of the five samples are shown in Figure 2. Pattern (a), (b), (c), (d), and (e) are from the layers annealed for 1, 2, 3, 4, and 5 h, respectively. Three distinctive diffraction peaks at $2\theta = 33.00$ $(\pm 0.05)^{\circ}$, 34.01 $(\pm 0.05)^{\circ}$, and 51.60 $(\pm 0.03)^{\circ}$ were identified as Gd_2O_3 (002), $Gd_2Zr_2O_7$ (004), and Ni-W (002) reflections, respectively. We observed an increase in relative intensity for the Gd₂Zr₂O₇ reflection when annealed for a longer time. As shown in Figure 2, the Gd₂Zr₂O₇ (004) reflection became four times more intense on annealing for 5 h compared to 1 h. As expected, an improvement of crystallinity, and therefore an improvement of grain growth, was obtained when the annealing time increased. Thus, a long annealing time of 4 to 5 h seemed to favor the grain growth in the Gd₂Zr₂O₇ layer. The relative intensity of the Gd₂O₃ (002) reflection remained similar in the five XRD patterns of Figure 2. No significant improvement in the crystallinity of GO layers was detected by XRD. One possible cause could be the limitation of the XRD technique for measuring such very thin layers.

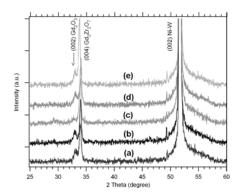


Fig. 3. XRD of electrodeposited $Gd_2O_3/Gd_2Zr_2O_7$ on Ni-W substrate after annealing in Ar:H₂ gas flow at (a) 800, (b) 950, (c) 1000, (d) 1050, and 1100 °C for 5 h.

Following the optimization of annealing time, we optimized the annealing temperature. The buffer layers were annealed at temperatures ranging from 800 to 1100 °C. Figure 3 shows the XRD patterns of $Gd_2O_3/Gd_2Zr_2O_7$ layers annealed for 5 h in Ar:H₂ gas at temperatures of (a) 800, (b) 950, (c) 1000, (d) 1050), and (e) 1100 °C, respectively. A significant variation in intensity was detected on the $Gd_2Zr_2O_7$ (004) reflection. We noted an increase of relative intensity, and thus an improvement in crystallinity and grain growth, when the temperature was increased from 800 to 1100 °C. No misorientation was detected by XRD, even for temperatures as low as 800 °C. $Gd_2O_3/Gd_2Zr_2O_7$ layers with the best crystallinity were obtained from a 5-h annealing at 1100 °C.

C. Electrodeposited $Gd_2O_3/Gd_2Zr_2O_7$ bilayer on Ni-W substrate

 Gd_2O_3 cap layers were electrodeposited on $Gd_2Zr_2O_7$ layers from 10 to 60 nm thick to determine the optimum cap layer thickness for YBCO growth. Surface morphology and texturing of $Gd_2O_3/Gd_2Zr_2O_7$ bilayers were analyzed for samples annealed at 1100 °C for 5 h. XRD patterns of the

 $Gd_2O_3/Gd_2Zr_2O_7$ layers are shown in Figure 4; (a), (b), (c), and (d) are from the Gd_2O_3 buffer layers of 10, 20, 30, and 60 nm thicknesses, respectively. Patterns (a) and (b) show a distinctive reflection at $2\theta = 33.98~(\pm~0.09)^\circ$, identified as a $Gd_2Zr_2O_7~(004)$ reflection. Patterns (c) and (d) show an additional XRD peak at $2\theta = 32.93~(\pm~0.03)^\circ$, which we identified as a $Gd_2O_3~(002)$ reflection. No misorientation was observed in the patterns. The grains in both Gd_2O_3 and $Gd_2Zr_2O_7$ layers were aligned along the c-axis, as shown by the presence of (00I) reflections only. In addition, analysis of the 2θ peak position revealed a shift of the $Gd_2Zr_2O_7~(004)$ reflection toward low angles, suggesting an increase of the d-spacing of $Gd_2Zr_2O_7$ when the thickness of Gd_2O_3 increased.

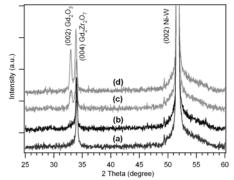


Fig. 4. X-ray diffraction of electrodeposited $Gd_2O_3/Gd_2Zr_2O_7$ on Ni-W substrate with various thicknesses of the GO layer: (a) 10, (b) 20, (c) 30, and (d) 60 nm after annealing in H_2 :Ar gas flow at 1100 °C for 5 h.

Surface morphologies of these films were analyzed by SEM. Micrographs (a) and (b) in Figure 5 show the surface of the annealed Gd₂Zr₂O₇ layer and Gd₂O₃/Gd₂Zr₂O₇ bilayer with a 30-nm-thick Gd₂O₃ cap layer, respectively. Both micrographs revealed smooth and crack-free surfaces with roughness of about 3 nm [5]. The Gd₂Zr₂O₇ layer shown in Figure 5(a) has more defects compared to the Gd₂O₃/Gd₂Zr₂O₇ layer in Figure 5(b). These defects are due primarily to the voids in the films, confirmed by TEM study of Gd₂Zr₂O₇ and Gd₂O₃/Gd₂Zr₂O₇ [11]. The electrodeposited Gd₂Zr₂O₇ layer had a void density as high as 10¹⁶ cm⁻². On the other hand, a TEM of the Gd₂O₃/Gd₂Zr₂O₇ stacking revealed a dense, crystalline, and textured Gd₂O₃ cap layer without any voids, which is ideal for the growth of good-quality YBCO films.

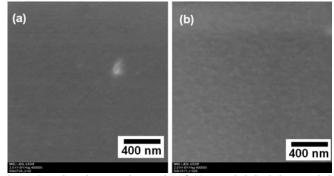


Fig. 5. Scanning electron micrographs of (a) annealed GZO layer and (b) GO/GZO layer with a 30-nm-thick GO cap on Ni-W substrate.

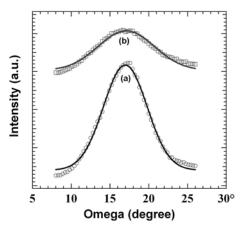


Fig. 6. Out-of-plane alignment of GO and GZO grains in electrodeposited GO/GZO layers annealed at 1100 °C for 5 h; omega scans of (a) GZO (004) reflection at 2θ = 33.95° and (b) GO (002) reflection at 2θ =32.93°.

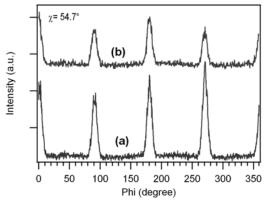


Fig.7. In-plane alignment of GO and GZO grains in a GO/GZO buffer structure annealed at 1100 °C for 5 h; phi scans of (a) GZO (222) reflection 2θ = 29.30° and (b) GO (111) reflection at 2θ =28.45°.

Out-of-plane and in-plane alignments of Gd₂Zr₂O₇ and Gd₂O₃ grains are shown in Figures 6 and 7, respectively. The rocking curves on the Gd₂Zr₂O₇ (004) and Gd₂O₃ (002) reflections show a mosaic spread along the c-axis of about 6° and 7.8° for Gd₂Zr₂O₇ and Gd₂O₃ grains, respectively. Compared to the Ni-W substrate (FWHM₁₁₁= 6.8°), the Gd₂Zr₂O₇ layer has an improved out-of-plane texture. On the other hand, the out-of-plane texture in Gd₂O₃ has been degraded. The in-plane texture in Figure 7 reveals a stacking of the Gd₂O₃ (111) crystal planes on the Gd₂Zr₂O₇ (222) crystal planes in the (a,b) plane of the substrate. The grain mosaic spreads in the (a,b) plane were 8.4° and 9.3° for Gd₂Zr₂O₇ and Gd₂O₃ layer, respectively. These values indicate a degradation of the texture compared to the Ni-W substrate (phi-scan, FWHM₁₁₁= 6.5°). Larger grain dispersion was observed for the Gd₂O₃ cap layer. In addition, we noticed that increasing the Gd₂O₃ cap layer thickness did not improve significantly the in-plane texture of Gd₂O₃.

IV. CONCLUSION

We obtained crystalline $Gd_2Zr_2O_7$ films with high-quality surface morphology by optimizing the ethanol concentration in the electrodeposition solution mixture. After annealing in Ar:H₂ gas mixture, crystalline $Gd_2Zr_2O_7$ films were obtained with a major orientation along <001> direction. The electrodeposited $Gd_2O_3/Gd_2Zr_2O_7$ bilayer annealed at 1100 °C for 5 h in an Ar:0.5%H₂ gas mixture yielded the best morphology and texture. Also, the $Gd_2O_3/Gd_2Zr_2O_7$ bilayer buffer architecture with a cap layer of about 20–30-nm-thick Gd_2O_3 was the best template.

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